

# Chemical Methodologies

## Supplementary File

### Lithium-Functionalization of Pyrrole-*n*-Carboxylic Acids (*n* = 1, 2, 3)

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This supplementary file includes the obtained results at the B3LYP/6-311+G\*\* theoretical level for all of investigated PnCA models for making comparison of different levels of calculations between 6-31+G\* and 6-311+G\*\* basis sets. The evaluated results have been supported by both theoretical levels with negligible differences; therefore, the results of 6-31+G\* basis set have been only listed in the original manuscript. Moreover, according to an earlier year, a new model has been studied here to see the stabilities of different types of Li-functionalized PnCA models. The H atom of carboxylic acid group of PnCA has been replaced by the Li atom to make the Li/H model (Fig. S1). Indeed, the main purpose of this work is keeping the initial carboxylic group and adding the extra Li/Li+. The values of  $E_{\text{Ads}}$  indicated that the Li/H model is more stable than Li-PnCA models but weaker than Li+-PnCA models. A potential reader could compare the results of this additional job with the results of an earlier work [23].

# Chemical Methodologies

**Table S1:** Molecular properties; obtained by the B3LYP/6-311+G\*\* theoretical level\*

Property	P1CA				P2CA				P3CA			
	Orig.	–Li	–Li+	Li/H	Orig.	–Li	–Li+	Li/H	Orig.	–Li	–Li+	Li/H
$d_{\text{C}_6\text{—O}_7}$ Å	1.35	1.32	1.32	1.27	1.36	1.35	1.33	1.28	1.36	1.33	1.33	1.28
$d_{\text{C}_6\text{—O}_8}$ Å	1.20	1.23	1.24	1.27	1.22	1.26	1.26	1.28	1.21	1.23	1.25	1.28
$d_{\text{O}_8\text{—Li}}$ Å	—	1.86	1.74	1.87	—	1.75	1.72	1.86	—	1.84	1.72	1.84
$\angle \text{O}_7\text{—C}_6\text{—O}_8^\circ$	124.32	124.73	122.31	122.70	122.61	120.00	120.67	121.00	121.79	122.10	120.00	120.55
$\angle \text{C}_6\text{—O}_8\text{—Li}^\circ$	—	130.70	176.28	82.11	—	171.40	172.73	82.93	—	139.16	178.60	82.84
$E_{\text{T}}$ keV	–10.85	–11.06	–11.05	–11.04	–10.85	–11.06	–11.05	–11.04	–10.85	–11.06	–11.05	–11.04
$\text{BSSE}$ eV	0.01	0.02	0.02	0.02	0.01	0.02	0.02	0.02	0.01	0.02	0.02	0.02
$E_{\text{Ads}}$ eV	—	–0.58	–1.91	–1.22	—	–0.66	–2.08	–1.61	—	–0.65	–2.27	–1.90
$I$ eV	6.54	2.98	10.10	5.90	6.61	2.79	10.31	5.97	6.54	2.59	10.07	5.95
$A$ eV	1.13	1.83	5.45	1.21	1.31	1.55	5.62	1.02	0.89	1.78	5.37	0.86
$E_{\text{G}}$ eV	5.41	1.15	4.65	4.69	5.30	1.24	4.69	4.95	5.65	0.81	4.70	5.09
$D_{\text{M}}$ Debye	1.53	4.043	11.40	4.80	1.45	1.26	11.45	3.77	3.554	8.24	10.41	1.69

\*See the models in Fig. (S1).

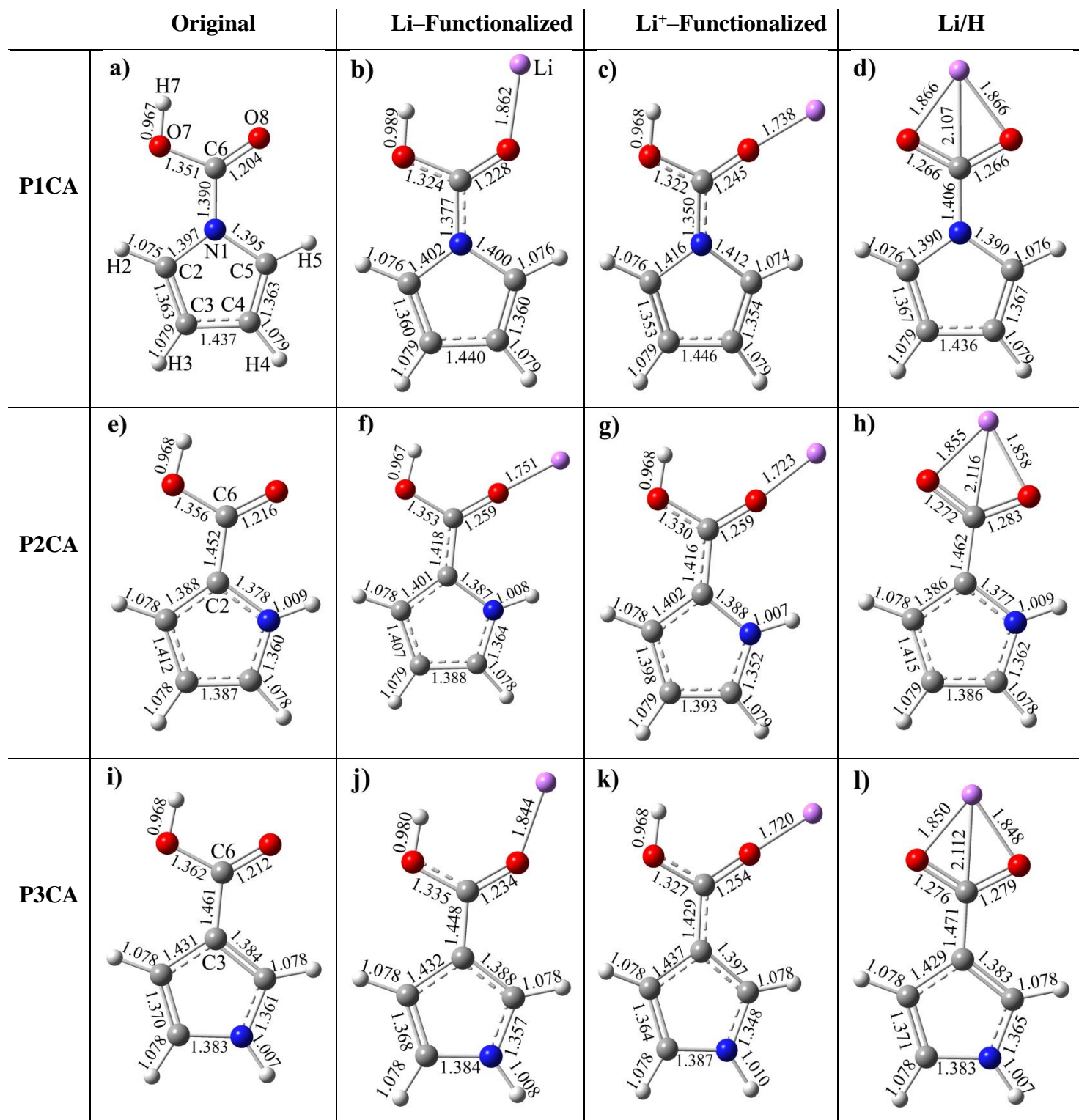
# Chemical Methodologies

**Table S2:** Atomic charges  $Q$ ; obtained by the B3LYP/6-311+G\*\* theoretical level\*

Atom	P1CA				P2CA				P3CA			
	Orig.	–Li	–Li+	Li/H	Orig.	–Li	–Li+	Li/H	Orig.	–Li	–Li+	Li/H
<b>N<sub>1</sub></b>	–0.46	–0.46	–0.44	–0.46	–0.50	–0.52	–0.50	–0.51	–0.52	–0.51	–0.49	–0.53
<b>C<sub>2</sub></b>	–0.03	–0.02	–0.02	–0.02	–0.01	–0.01	–0.03	0.02	0.03	0.04	0.06	0.01
<b>C<sub>3</sub></b>	–0.26	–0.25	–0.22	–0.28	–0.22	–0.22	–0.15	–0.22	–0.24	–0.25	–0.26	–0.23
<b>C<sub>4</sub></b>	–0.26	–0.26	–0.21	–0.28	–0.29	–0.29	–0.27	–0.30	–0.25	–0.24	–0.22	–0.25
<b>C<sub>5</sub></b>	–0.02	–0.18	–0.03	–0.02	–0.01	–0.01	0.06	–0.02	–0.05	–0.04	–0.02	–0.06
<b>C<sub>6</sub></b>	0.94	0.98	1.00	0.91	0.76	0.73	0.81	0.74	0.78	0.83	0.85	0.76
<b>O<sub>7</sub></b>	–0.69	–0.66	–0.63	–0.81	–0.68	–0.67	–0.63	–0.81	–0.69	–0.66	–0.62	–0.82
<b>O<sub>8</sub></b>	–0.60	–0.74	–0.86	–0.81	–0.63	–0.86	–0.89	–0.84	–0.62	–0.75	–0.87	–0.83
<b>Li</b>	—	0.04	0.96	0.90	—	0.28	0.96	0.89	—	0.01	0.96	0.88
<b>H<sub>1</sub></b>	—	—	—	—	0.43	0.42	0.42	0.42	0.41	0.41	0.43	0.40
<b>H<sub>2</sub></b>	0.23	0.23	0.24	0.23	—	—	—	—	0.22	0.23	0.22	0.22
<b>H<sub>3</sub></b>	0.22	0.22	0.24	0.21	0.23	0.23	0.24	0.23	—	—	—	—
<b>H<sub>4</sub></b>	0.22	0.22	0.24	0.21	0.22	0.22	0.24	0.21	0.23	0.24	0.25	0.23
<b>H<sub>5</sub></b>	0.23	0.23	0.22	0.23	0.21	0.21	0.23	0.20	0.21	0.21	0.23	0.20
<b>H<sub>7</sub></b>	0.50	0.48	0.51	—	0.48	0.48	0.42	—	0.48	0.48	0.50	—

\*See the models in Fig. (S1).

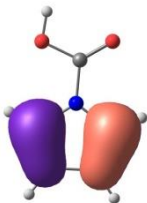
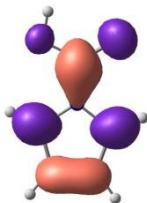
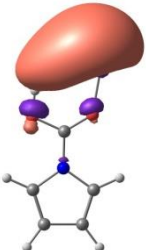
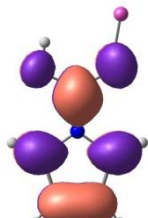
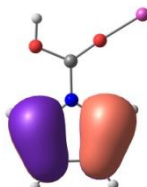
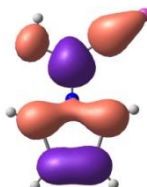
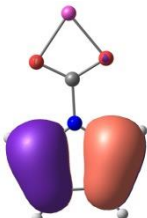
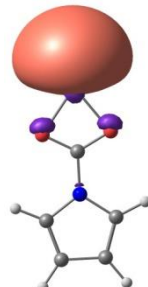
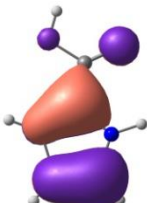
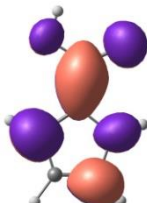
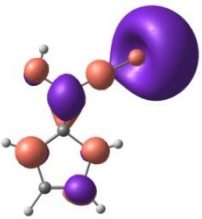
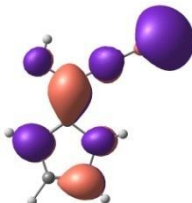
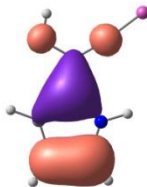
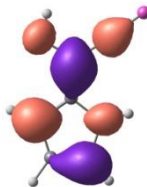
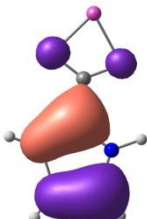
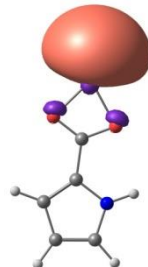
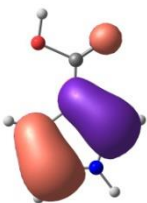
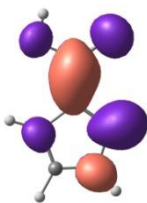
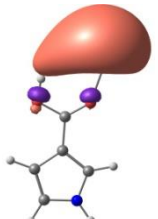
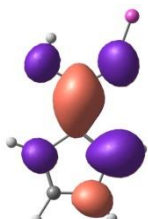
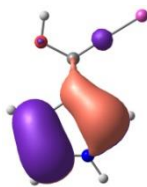
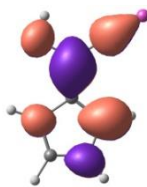
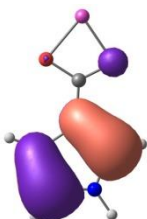
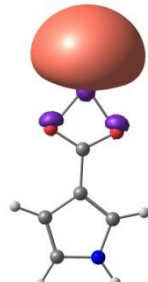
# Chemical Methodologies



**Figure S1:** The B3LYP/6-311+G\*\* optimized structures of original and singular Li/Li<sup>+</sup>-Functionalized and Li/H PnCA (n=1–3) models. The H atom of carboxylic acid group is replaced by the Li atom in the Li/H model.

# Chemical Methodologies

# Chemical Methodologies

	Original		Li-Functionalized		Li <sup>+</sup> -Functionalized		Li/H	
	HOMO	LUMO	HOMO	LUMO	HOMO	LUMO	HOMO	LUMO
P1CA								
P2CA								
P3CA								

**Figure S2:** The B3LYP/6-311+G\*\* HOMO and LUMO distribution patterns of original and singular Li/Li<sup>+</sup>-Functionalized and Li/H PnCA (n=1–3) models. The H atom of carboxylic acid group is replaced by the Li atom in the Li/H model.

# Chemical Methodologies